- 5. A. I. Ermakov and N. P. Yarosh, Tr. Prik. Bot., Genet. Selekts., <u>56</u>, No. 3, 3 (1976).
- 6. L. A. Baratova, N. A. Bokova, A. K. Yus'kovich, and I. A. Samylina, Khim. Prir. Soedin., 248 (1982).
- 7. A. U. Umarov, A. P. Markman, and M. T. Kisapova, Maslob-Zhir. Promst., 5, 12 (1971).

## COUMARINS OF Prunella vulgaris

S. I. Dmitruk

UDC 633.88:582.675.1:615.43

In a study of the epigeal part of <u>Prunella vulgaris</u> L. Family Lamiaceae collected in the environs of Tomsk in August, 1984, six benzo- $\alpha$ -pyrone derivatives have been detected by two-dimensional paper and thin-layer chromatography.

To isolate the coumarins, the comminuted herb was extracted exhaustively with hot 70% ethanol, the extract was concentrated and diluted with water (1:2) and was treated successively with petroleum ether (I) with chloroform (II), and with chloroform—ethyl acetate (2:1) (III) [1]. The combined dry residues II and III were transferred to a column of Woelm polyamide (5 × 20 cm). The fractions eluted by chloroform and chloroform—ethanol (2:1) were rechromatographed on Silufol plates in the benzene—ethyl acetate (2:1) system [2]. Two individual substances were isolated: A with  $R_{\rm f}$  0.72 and B with  $R_{\rm f}$  0.36. When the column was washed further with chloroform—acetone (7:3), the fractions contained a substance C with  $R_{\rm f}$  0.09 [3, 4]. The substances obtained were subjected to repeated recrystallization from methanol.

Substance A formed colorless acicular crystals with mp 233-236°C. UV spectrum:  $\lambda^{\rm C_2H_5OH}_{\rm max}$  325, 256 nm. The IR spectrum (tablets with KBr) had absorption bands at 1725 cm<sup>-1</sup> ( $\gamma$ -pyrone) and 3300 cm<sup>-1</sup> (OH group). On the basis of the results obtained, substance A was identified as umbelliferone.

Substance B formed white crystals with mp 205-207°C. On the basis of UV and IR spectroscopy and a mixed melting point it was identified as scopoletin.

Substance C formed yellowish acicular crystals with mp 269-271°C. UV spectrum:  $\lambda_{\text{max}}^{\text{C}_2\text{H}_5\text{OH}}$  262, 306, 355 nm. IR spectrum,  $\nu_{\text{max}}^{\text{KBr}}$  (cm<sup>-1</sup>): 1598, 1630, 1686 (C=C); 1718 (-C=O); and 3400 (OH group). The results of the investigation agreed with literature information for esculetin [5-7].

This is the first time that hydroxycoumarins have been isolated from Prunella vulgaris L.

## LITERATURE CITED

- V. P. Georgievskii, N. A. Kazarinov, and M. O. Karryev, Physicochemical Methods of Analyzing Biologically Active Substances of Plant Origin [in Russian], Ashkhabad (1976), p. 236.
- 2. L. I. Dranik and A. P. Prokopenko, Abstracts of Lectures at the Second Congress of Pharmacists of the UkrSSR [in Russian], Kiev (1972), p. 751.
- 3. M. V. Artem'eva, G. K. Nikonov, and M. O. Karryev, Khim. Prir. Soedin., 493 (1973).
- 4. M. E. Perel'son, Yu. I. Sheinker, and A. A. Savina, Spectra and Structures of Coumarins, Chromones, and Xanthones [in Russian], Moscow (1975), p. 223.
- 5. L. I. Kosheleva and G. K. Nikonov, Farmatsiya, 78-88 (1969).
- 6. G. A. Kuznetsova, Natural Coumarins and Furocoumarins [in Russian], Leningrad (1967).
- 7. A. A. Ponomarenko, N. F. Komissarenko, and K. A. Stukkei, Khim. Prir. Soedin., 661 (1971).

Tomsk State Medical Institute. Translated from Khimiya Prirodnykh Soedinenii, No. 4, pp. 510-511, July-August, 1986. Original article submitted January 15, 1986.